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[Group 13: The Boron Family - Chemistry LibreTexts](#)
The boron family contains elements in group 13 of the periodic table and include the semi-metal boron (B) and the metals aluminum (Al), gallium (Ga), indium (In), and thallium (Tl).

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The main group elements have always been used in this manner. Depending on the use of the elements, the resulting work can be described under any heading of chemistry. The group 13 elements have been special in this regard due to the very unique characters of the constituent elements. Thus, there is a dramatic change in the properties of the elements when proceeding through the series, B, Al.

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Boron is the first element of Group 13 and is the only metalloid of the group. Its chemical symbol is B, and it has an atomic number of 5. Boron has the electron configuration [He] 2s² 2p¹ and prefers an oxidation state of +3. Boron has no natural elemental form; it forms compounds which are

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series, B, A1

Group 13 Elements | Oxidation State, Chemical Reactivity ...

Chemistry Articles, Group 13 Elements Oxidation State And Chemical Reactivity, Group 13 Elements - Oxidation State and Chemical Reactivity . Boron is the first element of group 13 and only has 2 shells. Because of its small size, the first three ionization enthalpies are very high. This prevents boron from forming +3 ions, and hence it forms only covalent compounds. As we move from boron to

General Chemistry/Chemistries of Various Elements/Group 13

Group 13 (IIIA) consists of boron, aluminum, gallium, indium, and thallium. Boron is a metalloid, but the other elements are metals. All elements in this group have three valence electrons. Boron is a metalloid, but the other elements are metals.

The Elements of Group 13 - lardbucket

This is The Elements of Group 13 , section 22.1 from the book Principles of General Chemistry (v. 1.0). For details on it (including licensing), click here. To understand the trends in properties and the reactivity of the group 13 elements. As reductants, the group 13 elements are less powerful

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